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Abstract (Basic): GB 2077264 A

ML-236B derivs. of formula (I) ring-closed lactones, salts and esters are new: where R is a gp. (II) or (III): (I) are cholesterol biosynthesis inhibitors and are thus useful in treating hypercholesterolaemia.

(I) are prepd. by enzymatic hydroxylation of ML-236B, or ML-236B carboxylic acid or its salt or ester (see US398140). Pref. the enzyme is provided by a strain of Mucor, Phizopus, zygorynchus, circinella, Actinomucor, Gongronella, Phyomyces, Mortierella, Pycnoporus, Rhizoctonia, Absidia, Cunninghamella, Syncephalastrum or Streptomyces. A partic. suitable microorganism is Mucor hiemalis F. hiemalis, which gives at least 90% conversion of ML-236B and its derivs.

Title Terms: DERIVATIVE; ML; CHOLESTEROL; BIOSYNTHESIS; INHIBIT; USEFUL; TREAT; HYPERCHOLESTEROLAEMIC

Derwent Class: B03; B05; D16

International Patent Class (Main): C12P-007/62

International Patent Class (Additional): A61K-031/22; A61K-031/365; C07C-059/90; C07C-069/03; C07C-069/22; C07D-309/30; C12P-017/06; C12R-001/46 File Segment: CPI

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# UK Patent Application GB CO 2 077 264 A

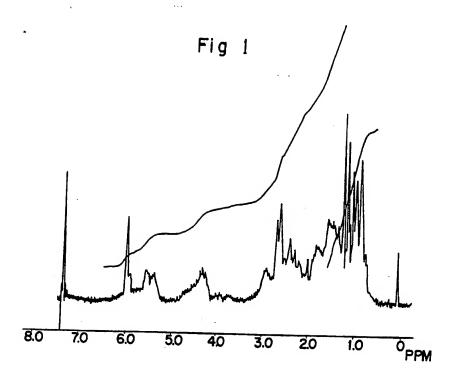
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- (58) Field of search C2C
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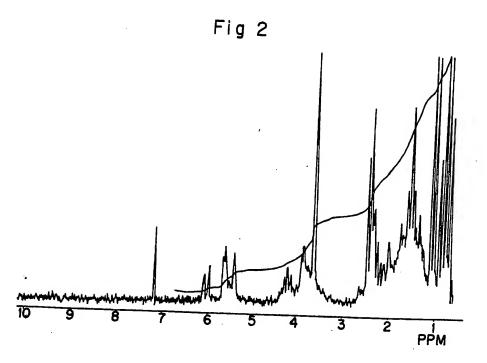
## (54) ML-236B derivatives and their preparation

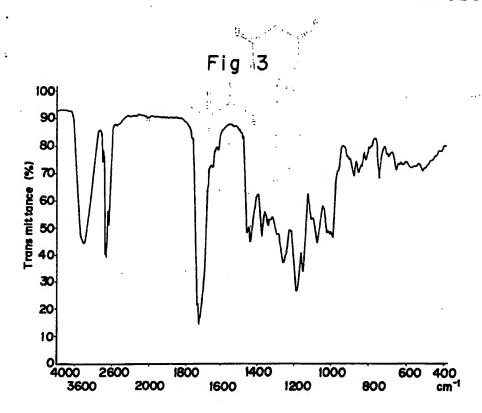
(57) Compounds of formula (I):

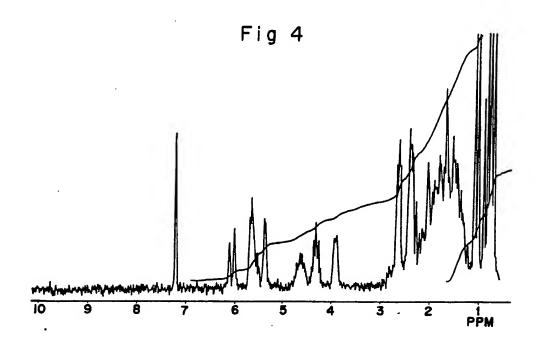
(wherein R represents a group of formula

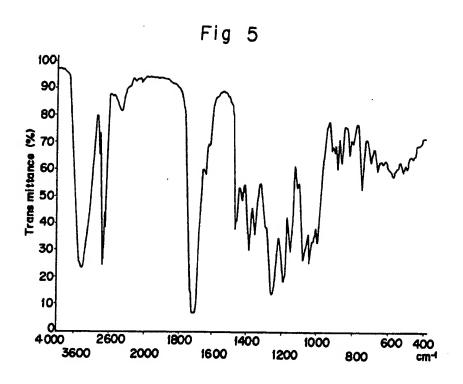
and the corresponding ring-closed lactones, salts (especially alkali metal salts) and esters (especially C1-C5 alkyl esters) thereof may be prepared by subjecting ML-236B, or ML-236B carboxylic acid or a salt or ester thereof to enzymatic hydroxylation, which may be effected by means of microorganisms of the genera Mucor, Rhizopus, Zygorynchus, Circinella, Actinomucor, Gongronella, Phycomyces, Martierella, Pycnoporus, Rhizoctonia, Absidia, Cunninghamella, Syncephalosporum and Streptomyces, or cell-free, enzymecontaining extracts from said microorganisms. The compounds are capable of inhibiting biosynthesis of cholesterol and are thus useful in the treatment of hypercholesteraemia.











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The present invention relates to a series of new derivatives of the known compound ML-236B, to processes for their preparation and to pharmaceutical compositions containing them.

ML-236B, which has the following chemical structure:

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is disclosed in U.S. Patent Specification No. 3,983,140. It has been isolated and purified from the metabolic products of microorganisms of the genus Penicillium, especially Penicillium citrinum, a species of blue mould. It has been shown to inhibit the biosynthesis of cholesterol by enzymes or 10 cultured cells separated from experimental animals by competing with the rate-limiting enzyme active in the biosynthesis of cholesterol, namely 3-hydroxy-3-methylglutaryl-coenzyme A reductase and, as a result, significantly reduces serum cholesterol levels of animals [Journal of Antibiotics, 29, 1346 (1976)]. A number of compounds structurally related to ML-236B have also been discovered and found to possess the ability to inhibit the biosynthesis of cholesterol.

We have now discovered a series of new compounds, which may be prepared by the enzymatic hydroxylation of ML-236B or of derivatives thereof, and which possess an ability to inhibit the biosynthesis of cholesterol which is at least comparable with, and in some instances substantially exceeds, that of ML-236B itself.

The compounds of the present invention are those hydroxycarboxylic acids of formula (I):

(in which R represents a group of formula

$$H_3$$
C  $CH_3$  or  $CH_3$   $CH_3$  OH

and ring-closed lact nes, salts and esters th reof.

The invention also provides a proc ss for preparing a compound of formula (I), r a ring-closed 25 lact ne, salt or est in thereof by the enzymatic hydroxylation of ML—236B, or ML—236B carboxylic acid, rasalt resterther of.

ML-236B carboxylic acid has the formula

One class of compounds of the present invention are those compounds of formula (II):

$$H_3$$
C  $H_3$   $H_0$   $CH_3$   $H_3$ 

5 (in which R¹ represents a hydrogen atom or a C<sub>1</sub>—C<sub>5</sub> alkyl group), pharmaceutically acceptable salts of the acid wherein R¹ represents a hydrogen atom, and the corresponding lactone of formula (III):

In view of the number of asymmetric carbon atoms in these compounds, a variety of geometric isomers are possible. Of these, the most important isomers are as follows:

(in which R1 is as defined above) and pharmaceutically acceptable salts of the acid wherein R1 represents a hydrogen atom, and the corresponding lactone of formula (V):

and compounds of formula (VI):

(in which  $R^1$  is as defined above), and pharmaceutically acceptable salts of the acid wherein  $R^1$  represents a hydrogen atom, and the corresponding lactone of formula (VII):

The hydroxy-carboxylic acid of formula (IV) in which R¹ represents a hydrogen atom is herein referred to as M—4 and derivatives of this acid, specifically the salts and esters, are named as derivatives of M—4, whilst the corresponding lactone of formula (V) is herein referred to as M—4 lactone. Similarly, the hydroxy-carboxylic acid of formula (VI) in which R¹ represents a hydrogen atom is referred to as M—4' and derivatives of this acid are referred to as derivatives of M—4', whilst the corresponding lactone of formula (VII) is referred to as M—4' lactone.

Another preferred class of compounds of the invention are those compounds of formula (VIII):

10 (in which R¹ is as defined above), and pharmaceutically acceptable salts of the acid in which R¹ 10 represents a hydrogen atom, and the corresponding lactone of formula (IX):

A variety of geometric isomers of these compounds are also possible, the most important being th following:

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(in which  $\mathsf{R}^1$  is as defined above), and pharmaceutically acceptable salts of the acid in which  $\mathsf{R}^1$  represents a hydrogen atom and the corresponding lactone of formula (XI):

5 and compounds of formula (XII):

(in which R¹ is as defined above), and pharmaceutically acceptable salts of the acid in which R¹ represents a hydrogen atom and the corresponding lactone of formula (XIII):

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The acid of formula (X) is herein referred to as IsoM-4 and its derivatives, such as salts and esters, are named as derivatives of IsoM-4, whilst the corresponding lactone of formula (XI) is herein referred to as IsoM-4 lactone. The acid of formula (XII) in which R1 represents a hydrogen atom is herein referred to IsoM-4', and its derivatives are named as derivatives of IsoM-4', whilst its corresponding lactone of formula (XIII) is herein referred to as IsoM-4' lactone.

Of the esters of the hydroxy-carboxylic acids of formula (I), the C<sub>1</sub>—C<sub>5</sub> alkyl esters are preferred. These alkyl groups may be straight or branched-chain groups and include, for example the methyl, ethyl, propyl, isopropyl, butyl and isobutyl groups, of which the methyl group is particularly preferred.

The hydroxy-carboxylic acids will also form salts with a variety of cations, particularly metals and most preferably alkali metals, such as sodium or potassium. The sodium salts are most preferred.

Of the compounds of the invention, the most preferred compounds are M-4 lactone, M-4 sodium salt, M-4 methyl ester, IsoM-4' lactone, IsoM-4' sodium salt and IsoM-4' methyl ester, -4 sodium salt being particularly preferred.

The compounds of the invention may be prepared by the enzymatic hydroxylation of ML-236B or 15 of a derivative thereof, specifically ML-236B carboxylic acid or a salt or ester thereof.

This enzymatic hydroxylation may be effected as part of the mammalian metabolism of -236B or a derivative thereof, for example by administering ML--236B to a suitable animal, collecting a metabolic product, e.g. urine, and then separating the desired compound or compounds of 20 20 the invention from this metabolic product. Alternatively, the liver or an enzyme-containing extract from the liver may be used instead of the living animal. However, processes employing the animal metabolism or animal products have a relatively low productivity and are difficult to carry out reproducibly. Accordingly, we prefer to employ microorganisms or enzyme- containing extracts from their microorganisms. 25

Accordingly, the process of the present invention is preferably effected using a microorganism capable of converting ML-236B or a derivative thereof to a compound of the present invention or using an enzyme-containing extract of such a microorganism. Particularly preferred microorganisms are those of the following genera: Mucor, Rhizopus, Zygorynchus, Circinella, Actinomucor, Gongronella, Phycomyces, Martierella, Pycnoporus, Rhizoctonia, Absidia, Cunninghamella, Syncephalastrum and

30 Streptomyces. In particular the following species are preferred: Absidia coerulea

Cunninghamella echinulata Syncephalastrum racemosum Streptomyces roseochromgenus

35 Mucor hiemalis f. hiemalis Mucor bacilliformis Mucor circinelloides f. circinelloides

Mucor hiemalis f. corticolus Mucor dimorphosporus

40 Mucor fragilis Mucor genevensis Mucor globosus Mucor circinelloides f. griseo-cyanus Mucor heterosporus

45 Mucor spinescens Rhizopus chinensis Rhizopus circinans Rhizopus arrhizus

Zygorynchus moelleri 50 Circinella muscae Circinella rigida

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111	Circinella umbellata	
••	Actinomucor elegans	
÷	Phycomyces blakesleeanus	
	Martierella isabellina	5
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	Pycnoporus Coccineus	
	Rhizoctonia solani	
	Syncephalastrum nigricans	
	Absidia glauca var. paradoxa	
10	Amongst strains of the above species, the following are particularly preferred:	10
	Absidia coerulea IFO—4423	
	Cunninghamella echinulata IFO—4445	
	Cunninghamella echinulata IFO—4444	
	Cunninghamella echinulata ATCC—9244	
15	Syncephalastrum racemosum IFO—4814	15
13	Syncephalastrum racemosum IFO—4828	•
	Streptomyces roseochromogenus NRRL—1233	
	Streptomyces roseochromogenus IFO—3363	
	Streptomyces roseochromogenus IFO—3411	
20	Mucor hiemalis f. hiemalis 1FO—5834	20
20	Mucor hiemalis I. hiemalis IFO—5884  Mucor hiemalis f. hiemalis IFO—5303	
	Mucor niemalis 1. niemalis 160—5303 Mucor hiemalis f. hiemalis 1F0—8567	
	Wideon riemais I. niemais Iro—6307	
	Mucor hiemalis f. hiemalis IFO —8449	
	Mucor hiemalis f. hiemalis IFO—8448	25
25	Mucor hiemalis f. hiemalis IFO—8565	
	Mucor hiemalis f. hiemalis CBS—117.08	
	Mucor hiemalis f. hiemalis CBS—109.19	
	Mucor hiemalis f. hiemalis CBS—200.28	
	Mucor hiemalis f. hiemalis CBS—242.35	30
30	Mucor hiemalis f. hiemalis CBS110.19	JU
	Mucor hiemalis f. hiemalis CBS—201.65	
	Mucor bacilliformis NRRL—2346	
	Mucor circinelloides f. circinelloides IFO—4554	
	Mucor circinelloides f. circinelloides IFO5775	
35	Mucor hiemalis f. corticolus NRRL—12473	35
	- Mucor dimorphosporus IFO—4556	
•	Mucor fragilis CBS—236.35	
	Mucor genevensis IFO4585	
•	Mucor globosus NRRL 12474	
40	Mucor circinelloides f. griseo-cyanus IFO—4563	40
	Mucor heterosporus NRRL—3154	
	Mucor spinescens IAM—6071	
	Rhizopus chinensis IFO4772	
	Rhizopus circinans ATCC—1225 .	
45	Rhizopus arrhizus ATCC—11145	45
40	Zygorynchus moelleri IFO—4833	
	Circinella muscae IFO—4457	
	Circinella rigida NRRL—2341	
	Circinella umbellata NRRL—1713	
50	Circinella umbellata IFO-4452	50
50	Circinella umbellata IFO—5842	
	Phycomyces blakesleeanus NRRL—12475	
	Martierella isabellina IFO—6739	
	Gongronella butleri IFO—8080	55
55	Pycnoporus coccineus NRRL—12476	33
	Rhizoctonia solani NRRL—12477	
	Syncephalastrum nigricans NRRL—12478	
•	Syncephalastrum nigricans NRRL—12479	
	Syncephalastrum nigricans NRRL—12480	~~
60	Absidia glauca var. paradoxa IFO4431	60
	Actinomucor elegans ATCC—6476	
	The microorganisms listed abovare available from International Culture Collections, as indicated	
	by the codes appended to their accession numbers, which codes have the following meanings.	
	IFO = Institute f r Fermentation, Osaka, Japan	
65	NRRL = Agricultural R search Culture C llection, Illin is, USA	65

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	CBS = Centraal Bureau voor Schimmelcultures, Netherlands	
	IAM = Institute f Applied Microbiology, Tokyo, Japan	٠
	ATCC = American Type Culture Coll ction, Maryland, USA.	
_	Of the spicies noted above, the following are particularly preferred:	_
5	, 1407-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1	5
	Cunninghamella echinulata	
	Syncephalastrum racemosum	
	Mucor hiemalis f. hiemalis Mucor bacilliformis	
10	Mucor circinelloides f. circinelloides	10
	Mucor hiemalis f. corticolus	
	Mucor dimorphosporus	
	Mucor fragilis	
	Mucor genevensis	
15	Mucor globosus	15
	Mucor circinelloides f. griseo-cyanus	
	Mucor heterosporus	
	Mucor spinescens	
	Pycnoporus coccineus	
20	Rhizoctonia solani	20
	Syncephalastrum nigricans	
	and the following are particularly preferred strains of the species:	
	Absidia coerulea IFO—4423	
	Cunninghamella echinulata IFO—4445	
25	Cunninghamella echinulata IFO—4444	25
	Cunninghamella echinulata ATCC—9244	
	Syncephalastrum racemosum IFO—4814	
	Syncephalastrum racemosum IFO—4828 Mucor hiemalis f. hiemalis IFO—5834	
30	Mucor hiemalis 1. hiemalis IFO—5834  Mucor hiemalis f. hiemalis IFO—5303	30
30	Mucor hiemalis f. hiemalis IFO—8567	30
	Mucor hiemalis f, hiemalis 1FO—8449	
	Mucor hiemalis f, hiemalis IFO—8448	
	Mucor hiemalis f. hiemalis IFO—8565	
35	Mucor hiemalis f, hiemalis CBS—117.08	35
	Mucor hiemalis f. hiemalis CBS—109.19	
	Mucor hiemalis f. hiemalis CBS—200.28	
	Mucor hiemalis f. hiemalis CBS—242.35	
	Mucor hiemalis f. hiemalis CBS—110.19	
40	Mucor hiemalis f. hiemalis CBS—201.65	40
	Mucor bacilliformis NRRL—2346	
	Mucor circinelloides f. circinelloides IFO—4554	
	Mucor circinelloides f. circinelloides IFO—5775	
4=	Mucor hiemalis f. corticolus NRRL—12473	
45	Mucor dimorphosporus IFO—4556	45
	Mucor fragilis CBS—236.35	
	Mucor genevensis IFO—4585 Mucor globosus NRR 12474	
•	Mucor circinelloides f. griseo-cyanus IFO—4563	
50	Mucor heterosporus NRRL—3154	50
50	Mucor spinescens IAM—6071	50
	Pycnoporus coccineus NRRL—12476	
	Rhizoctonia solani NRRL—12477	
	Syncephalastrum nigricans NRRL—12478	
55	Syncephalastrum nigricans NRRL—12479	<b>5</b> 5
	Syncephalastrum nigricans NRRL—12480	
	For the preparation of compounds of formulae (IV) and (V) and their salts, the following species are	
	preferred:	,
•	Mucor hiemalis f. hiemali	
60	Mucor circinelloides f. circinelloides	60
	Mucor fragilis	
	Mucor genevensis	
	Mucor circinelloides f. griseo-cyanus	
65	Pycnoporus coccineus Rhizoctonia solani.	ė-
UÜ	rrinzucturia Sulaili.	65

•	Apple of the control		
	Forth preparation o	f compounds of formula (VI) and (VIII) and their salts, the species and Sympophalastrum racemosum are preferred.	
	For the preparation o	f compounds of formula (VIII) and (IX) and their saits, the speci is Absidia	
	anorulas and Cunningham	ella echinulata are preferred. isted above, Mucor hiemalis f. hiemalis is particularly preferred since it is able	5
:5	to convert ML-236B and	its derivatives to the desired compounds of formula (I) at a conversion of	J
	90% or even higher.	236B or derivatives thereof to compounds of formula (I) may be achieved by	
	contacting the complete of	allular microorganism or, in some cases, a cell-free extract from the	
10	missopranism with MI	2368 or a derivative thereof. The form of the compound produced will	10
. •	demand upon the culture of	anditions and the form of microorganism employed. Thus, for example, if the	
	complete cellular microord	anism is cultivated in the presence of ML—2308 or a derivative thereof, the	
	product will be the carboxy	lic acid, the lactone or alkali metal salt, depending upon the culture pH. On the other hand, if the ML—236B or derivative thereof is simply	
15	conditions, particularly the	illular system or with a cell-free extract, the compound of the invention is	15
	absoluted in the form of an :	alkali metal sait.	
	The progress of the c	onversion reaction may be determined by assaying samples of the reaction	
	of M / lactone	of the reaction to determine the degree of conversion. For example, the may be assayed by liquid chromatography employing as a carried Micro	
20	Bondanak C (manufactur	ed by Waters Co. USA) and as the solvent 62% y/v aqueous methanol at the	20
	mote of 1 mil/minute \M/hen	detected using its ultraviolat absorption at 23 / nm, ivi—4 gives a peak at a	
	retention time of 10 minut	described disting to the assay. Similar techniques are available for	
	assaying the other compou	nisms are to be cultivated in the presence of ML2368 or a derivative	
25	thereof to product the com	nounds of the invention, the culture conditions and culture media employed	25
	will be chosen having rega	rd to the narticular microorganism to be cultivated. Since the species of	
	microorganism proposed to	or use in the process of the present invention are well known, culture lia for use with these microorganisms are also well known.	
	The compounds of th	e invention may be separated from the reaction mixture by conventional	
30	means for example by filte	ring off microbial cells (if necessary) and then subjecting the remaining	30
	mixture to any combination	of thin layer chromatography, column chromatography or high performance avarious compounds of the invention, where two or more are prepared	
	liquid chromatography. The	d from each other in the course of one or more of these chromatographic	
	purification steps.		
35	In addition to the con	npounds of the invention, there may, in some cases, also be prepared a	35
	compound which we have	designated M—3 and which is known under the name 3',5'-dihydroxy- copending application entitled "Hydronaphthalene Derivatives, their	
	Preparation and Use". This	may also be separated in the same way.	
	We have found that t	he compounds of the invention give a 50% inhibition of cholesterol	40
40	biosynthesis at concentrations	ons comparable with, or, in some cases, significantly less than, the ML—236B and certain other similar known compounds. The inhibitory	40
	activities of certain of the o	compounds of the invention, in terms of the concentration in $\mu g/m$ required	
	to inhibit cholesterol biosys	nthesis by 50% [measured by the method described in the Journal of	
	Biological Chemistry, 234,	2835 (1959)] are as follows:	
45	M-4 methyl ester	0.001	45
	M—4 sodium salt	0.0008	
	M—4 lactone	0.016	
	IsoM—4' methyl ester	0.007	
	IsoM—4' lactone	0.013	
50	M41	0.019	50
	M—41 sodium salt	0.00049	
	ML236B	0.01	

The inventi in is further illustrated by the following Examples.

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EX/	ΔМ	Pi	F	1
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Preparation of M-4 lactone

Twenty 500 ml Sakaguchi flasks, each containing 100 ml of a medium having the comp sition described below, were inoculated with spores of *Absidia coerulea* IFO 4423. The flasks were subject d to shaking culture at 26°C and 120 strokes per minute (s.p.m.) for 2 days. At the nd of this time, the sodium salt of ML—236B was added to each of the flasks to a final concentration of 0.05% w/v. Cultivation was continued at 26°C and 120 s.p.m. for a further 5 days.

The composition of the medium was (percentages are w/v):

	Glucose	2.0%	
10	K₂HPO₄	0.15%	10
	MgSO <sub>4</sub> .7H <sub>2</sub> O	0.15%	
	NH₄NO₃	0.1%	
	Peptone	0.1%	
	Corn steep liquor	0.2%	
15	Yeast extract	0.1%	15
	ZnSO <sub>4</sub> .7H <sub>2</sub> O	0.001%	
	Tap water	the balance (adjusted to pH 7.0).	

After completion of the cultivation, the reaction liquor was filtered, and the filtrate was adjusted 20 with trifluoroacetic acid to pH 3. The resulting mixture was extracted with three 1 litre portions of ethyl acetate, to give extracts containing M-4. This compound shows an Rf value of 0.45 on thin layer chromatography (TLC) (Plate: Merck silica gel Art 5715; solvent: a 50:50:3 by volume mixture of benzene, acetone and acetic acid). The combined extracts were washed with saturated aqueous sodium chloride, and then a catalytic amount of trifluoroacetic acid was added for lactonization. The resulting mixture was then washed with a 1% w/v aqueous solution of sodium bicarbonate, dried over anhydrous 25 sodiums sulphate and evaporated under reduced pressure to dryness. The residue was subjected to preparative liquid chromatography, System 500 using a Prep PAK-500/C<sub>18</sub> cartridge manufactured by Waters Associates (Prep PAK is a Trade Mark). Purification with a 55% v/v aqueous methanol system yielded 50.1 mg of M-4 lactone. 30 30 M-4 lactone has the following physical properties. 1) Nuclear Magnetic Resonance Spectrum: The NMR spectrum measured at 60 MHz in deuterochloroform using tetramethylsilane as the internal standard is shown in Figure 1 of the accompanying drawings. 2) Ultraviolet absorption spectrum (methanol solution)  $\lambda_{max}$ nm: 230; 236.7; 244.6. 35 3) Infrared absorption spectrum (liquid film) v cm<sup>-1</sup>: 3400, 2950, 1725. 35 4) Thin layer chromatography:

4) Thin layer chromatography:
TLC plate: Merck silica gel Art 5715;
Solvent: benzene, acetone, acetic acid (50:50:3 by volume);

Solvent: benzene, acetone, acetic acid (50:50:3 by volume); Rf value: 0.62.

40 EXAMPLE 2

48 mg of M—4 lectone were prepared following the same procedures as in Example 1, but using

48 mg of M—4 lactone were prepared following the same procedures as in Example 1, but using Cunninghamella echinulata IFO 4445.

EXAMPLE 3
30 mg of M—4 lactone were prepared following the same procedures as in Example 1, but using
45 Streptomyces roseochromogenus NRRL 1233.
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EXAMPLE 4
5 mg of M—4 lactone were prepared following the same procedures as in Example 1, but using Syncephalastrum racemosum IFO 4814.

EXAMPLE 5
50 6 mg of M—4 lactone were prepared following the same procedures as in Example 1, but using 50
Syncephalastrum racemosum IFO 4828.

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### EXAMPLE 6

Preparation of IsoM—4' methyl ester Twenty 500 ml Sakaguchi flasks, each containing 100 ml of a m dium having the composition described below, were inoculated with spores of Absidia coerulea IFO 4423. The flasks were subjected 5 to shaking culture at 120 s.p.m. and 26°C for 2 days. At the end of this time, the sodium salt of ML-236B was added to each of the flasks to a final concentration of 0.05% w/v. Cultivation was continued at 120 s.p.m. and 26°C for a further 5 days.

The composition of the medium was (percentages are w/v):

	Glucose	2.0%.	
10	K₂HPO₄	0.15%	10
	MgSO <sub>4</sub> .7H <sub>2</sub> O	0.15%	
	NH <sub>4</sub> NO <sub>3</sub>	0.1%	
	Peptone	0.1%	
	Corn steep liquor	0.2%	
15·	Yeast extract	0.1%	15
	ZnSO <sub>4</sub> .7H <sub>2</sub> O	0.001%	
	Tap water	the balance (adjusted to pH 7.0).	

After completion of the cultivation, the reaction liquor was filtered, and the filtrate was adjusted 20 with trifluoroacetic acid to pH 3. The resulting mixture was extracted with three 1 litre portions of ethyl acetate to give extracts containing IsoM-4'. This compound has an Rf value of 0.45 on thin layer chromatography (plate: Merck silica gel Art 5715; solvent: a 50:50:3 by volume mixture of benzene, acetone and acetic acid). The extract was washed with a saturated aqueous solution of sodium chloride, and then an ethereal solution of diazomethane was added. The mixture was allowed to stand for 30 25 25 minutes and then evaporated under reduced pressure to dryness. The residue was placed on a Lobar

column (Merck Si 60, Size A) and purified using as the solvent system a 1:1 by volume mixture of benzene and ethyl acetate. There were obtained 200 mg of an IsoM-4' methyl ester fraction. This fraction was further purified on a Lobar column (Merck RP-8, Size A) using 35% v/v aqueous acetonitrile as the eluent to give 78 mg of pure IsoM—4' methyl ester, having the following

characteristics:

35

40

1) Nuclear Magnetic Resonance Spectrum:

The NMR spectrum measured at 100 MHz in deuterochloroform using tetramethylsilane as the internal standard is shown in Figure 2 of the accompanying drawings.

2) Mass spectrum:

Measurement was made [after silylation with N,O-bis(trimethylsilyl)trifluoroacetamide] using a mass spectrometer, type D-300 manufactured by Nippon Electronics.

M/e: 654 (M+), 552, 462, 372, 272, 233, 231.

3) Ultraviolet absorption spectrum (methanol solution)  $\lambda_{max}$ nm: 229, 234.8; 244.5.

4) Infrared absorption spectrum (liquid film): As shown in Figure 3 of the accompanying drawings. 40

5) Thin layer chromatography:

TLC plate: Merck silica gel Art 5715;

Solvent: benzene, acetone (1:1 by volume);

Rf value: 0.88.

By operating as described above but replacing the diazomethane by another appropriate diazoalkane, it is possible to produce other esters of IsoM-4'.

**EXAMPLE 7** 

Preparation of IsoM-4' lactone

The procedure described in Example 6 was r peated up to and including xtraction with thyl acetate to give xtracts containing IsoM-4'. The combined extracts were wash d with a saturated aqueous solution of sodium chlorid and then evap rated to dryness to give the lacton product. Th resulting r sidue was placed on a Lobar column (M rck Si 60, Siz A) and purified using as the solv nt system a 1:1 by v lum mixture of benzene and ethyl acetate, to afford 198 mg of IsoM-4' lact n.

standard.

This product was furth r purified by means of a Lobar column (M rck RP-8, Size A) eluted with 35% v/v aqueous acetonitrile, to giv 82 mg of pure IsoM-4 lactone, having the following charact ristics: 1) Nucl ar Magnetic Resonance Spectrum: The NMR spectrum measured at 100 MHz in deuterochloroform using tetramethylsilane as th 5 5 Internal standard is shown in Figure 4 of the accompanying drawings. 2) Ultraviolet absorption spectrum (methanol solution)  $\lambda_{\max}$ nm: 229, 234.8; 244.5. 3) Infrared absorption spectrum (liquid film): As shown in Figure 5 of the accompanying drawings. **EXAMPLE 8** 63 mg of IsoM-4' lactone were prepared, following the same procedures as in Example 7, but 10 10 using Cunninghamella echinulata IFO 4445. 24 mg of IsoM---4' lactone were prepared, following the same procedures as in Example 7, but using Syncephalastrum racemosum IFO 4814. 35 mg of IsoM-4' lactone were prepared, following the same procedures as in Example 7, but 15 15 using Syncephalastrum racemosum IFO 4828. **EXAMPLE 11** 12 mg of IsoM-4' lactone were produced according to the process described in Example 7, but using Streptomyces roseochromogenus NRRL 1233. 20 20 EXAMPLE 12 Preparation of IsoM---4' sodium salt In a small amount of acetone were dissolved 10 mg of IsoM-4' lactone. To the solution was added an equivalent amount of sodium hydroxide and the mixture was allowed to stand for 1 hour. The pH of the resulting mixture was adjusted with 0.1N hydrochloric acid to a value of 8.0. The acetone was 25 then distilled off, and the residue was placed on an XAD-20 column (about 20 ml). The column was 25 washed with distilled water and then eluted with 50 ml of 50% v/v aqueous acetone. The acetone was again distilled off, and the residue was freeze-dried to afford 6 mg of IsoM-4' sodium salt, having the following characteristics: 1) Ultraviolet absorption spectrum (methanol solution)  $\lambda_{max}$ nm: 229 (shoulder); 235, 245 30 (shoulder). 30 Infrared absorption spectrum (KBr) p cm<sup>-1</sup>: 3400, 2850, 1710, 1580. 3) Thin layer chromatography: TLC plate: Merck silica gel Art 5715; Solvent: benzene, acetone, acetic acid (50:50:3 by volume); 35 Rf value: 0.45. 35 **EXAMPLE 13** Preparation of M-4 methyl ester Twenty 500 ml Sakaguchi flasks, each containing 100 ml of a medium of the same composition as shown in Example 1, were inoculated with spores of Absidia coerulea IFO 4423. The flasks were 40 subjected to shaking culture at 26°C and 120 s.p.m. for 2 days. The sodium salt of ML-236B was then 40 added to each of the flasks to a final concentration of 0.05 w/v. Cultivation was continued at 26°C and 120 s.p.m. for a further 5 days. After completion of the cultivation, the reaction liquor was filtered, and the filtrate was adjusted with trifluoroacetic acid to pH 3. The resulting mixture was extracted with three 1 litre portions of ethyl 45 acetate, to give extracts containing M-3, M-4 and IsoM-4'. Both M-4 and IsoM-4' show an Rf 45 value of 0.45 on thin layer chromatography (Plate: Merck silica gel Art 5715; solvent: a 50:50:3 by volume mixture of benzene, acetone and acetic acid). The combined extracts were washed with saturated aqueous sodium chloride, and then an ethereal solution of diazomethane was added. The mixture was allowed to stand for 30 minutes and then evaporated under reduced pressure to dryness. 50 When the residue was placed on a Lobar column (Merck Si 60, Size A), and purification was effected 50 using a 1:1 by volume mixture of benzene and ethyl acetate, a fraction containing IsoM-4' methyl ester and a fraction containing M—4 methyl ester were separated. There were obtained 185.3 mg of the latter active fraction, from which 20 mg of pure M-4 methyl ester were obtained as a colourless oil by using a Lobar column (Merck RP-8, Size A) and eluting with 35% v/v aqueous acetonitrile. M-4 methyl st r has th following characteristics: 55 1) Nuclear Magnetic Resonance Spectrum: M asurement was made at 200 MHz in d uterochlorof rm using tetram thylsilan as the internal

```
\delta ppm; \delta ppm; \delta and \delta is a sequence of the sequence \delta and \delta is a sequence 
0.89 (3H, doublet, J = 6.5, Hz); 3 170(12)
           1.12 (3H, d ublet, J = 6.8 Hz);
                                                                         The second of the setting of the
                                                                                                                                                                                                     5
5 1.1—1.7 (10H, multiplet);
           2.34 (1H, sextuplet, J = 7 Hz);
          2.3—2.5 (2H, multiplet);
           2.49 (2H, doublet, J = 6.4 \text{ Hz});
           2.58 (1H, multiplet);
                                                                                                                                                                                                   10
           3.72 (3H, singlet);
           3.78 (1H, multiplet);
           4.25 (1H, quintet, J = 7Hz);
           4.4 (1H, multiplet);
           5.42 (1H, multiplet):
                                                                                                                                                                                                   15
          5.56 (1H, multiplet);
           5.90 (1H, doubled doublet, J = 9.8 and 5.6 Hz);
           5.99 (1H, doublet, J = 9.8 Hz).
                     2). Mass spectrum:
                     Measurement was made [after silylation with N,O-bis(trimethylsilyl)trifluoroacetamide] using a
                                                                                                                                                                                                   20
  20 mass spectrometer, type D—300 manufactured by Nippon Electronics.
                                                        M/e: 654(M+), 552, 462, 372, 290, 272, 233, 231.
                     3) Ultraviolet absorption spectrum (ethanol solution) \lambda_{max}nm: 230.1; 237.3; 246.4.
                     4) Infrared absorption spectrum (liquid film) v \text{ cm}^{-1}: 3400, 2950, 1730.
                     5) Thin layer chromatography:
                                                                                                                                                                                                   25
                    TLC plate: Merck silica gel Art 5715;
  25
                     Solvent: benzene and acetone (1:1 by volume);
                     Rf value: 0.88.
                     By operating as described above but replacing the diazomethane by another appropriate
           diazoalkane, it is possible to produce other esters of M-4.
                                                                                                                                                                                                    30
  30 EXAMPLE 14
           Preparation of Sodium Salts of M—4 and IsoM—4'
                    The procedure described in Example 1 was repeated (except that the culture medium contained
           Na<sub>2</sub>HPO<sub>4</sub> instead of K<sub>2</sub>HPO<sub>4</sub>) up to and including filtration of the reaction liquor. The filtrate was then
           adsorbed on an HP-20 column (manufactured by Mitsubishi Chemical Industries). After washing the
   35 column with water, fractions containing M—4 sodium salt, IsoM—4' sodium salt and M—3 sodium
                                                                                                                                                                                                    35
           salt were eluted with 50% v/v aqueous acetone. The active fractions were freeze-dried, giving 830 mg
           of a freeze-dried product, which was purified by repeatedly subjecting it to high-performance liquid
           chromatography (column: Micro Bondapak C<sub>18</sub>, 40% v/v aqueous methanol 1 ml/min.) to give 32 mg of
                  -4 sodium salt and 280 mg of IsoM---4' sodium salt.
                    The properties of the IsoM-4' sodium salt were identical to those of the product of Example 12
                                                                                                                                                                                                    40
           and the properties of the M-4 sodium salt are as follows:
                     1) Nuclear Magnetic Resonance Spectrum:
                     Measurement was made at 200 MHz in deuteromethanol using tetramethylsilane as the internal
           standard.
                                                                                                                                                                                                    45
   45 δ ppm:
           0.91 (3H, triplet, J = 7.5 \text{ Hz});
           0.92 (3H, doublet, J = 7 \text{ Hz});
            1.12 (3H, doublet, J = 7 Hz);
            1.1—1.8 (10H, multiplet);
                                                                                                                                                                                                    50
    50 2.25 (1H, doubled doublet, J = 15 and 7.6 Hz);
           2.34 (1H, doubled doublet, J = 15 and 5.5 Hz):
           2.2-2.4 (3H, multiplet);
           2.48 (1H, multiplet);
           3.68 (1H, multiplet);
                                                                                                                                                                                                    55
   55. 4.07 (1H, multiplet);
            4.28 (1H, multiplet);
           5.36 (1H, multiplet);
            5.48 (1H, doubled doublet; J = 3 and 2 Hz);
            5.88 (1H, doubled d ublet, J = 9.6 and 5.3 Hz);
                                                                                                                                                                                                    60
   60 5.98 (1H, doublet, J = 9.8 Hz).
                      2) Ultraviolet abs rption sp ctrum (methanol solution) \lambda_{max}nm: 230.0; 237.2; 245.0.
                     3) Infrared absorption spectrum (KBr) v cm<sup>-1</sup>: 3400, 2900, 1725, 1580.
```

	4) Thin layer chr TLC plate: Mercl Solvent: benzen: Rf value: 0.45.	omatography: c silica gel Art 5715; e, acetone and ac tic acid (50:50:3 by volu	me);	•
5	EXAMPLE 15 18 mg of M4 using Cunninghamella	methyl ester were prepared, following the s echinulata IFO 4445.	ame procedures as in Example 13, but	5
10		methyl ester were prepared, following the s eseochromogenus NRRL 1233.	ame procedures as in Example 13, but	10
		methyl ester were prepared, following the s n racemosum IFO 4814.	ame procedures as in Example 13, but	
15		methyl ester were prepared, following the san racemosum IFO 4828.	ame procedures as in Example 13, but	15
20	mg/kg/day and their ur	le, average weight 10 kg) were administere ine was collected for 3 days. 3 litres of colle	ected urine were passed through a	20
25	500 ml XAD—2 column, eluted with 500 ml of 50% v/v aqueous acetone, and, after distilling off the acetone under reduced pressure, the residual liquid was adjusted to pH 3 by the addition of trifluoroacetic acid. The mixture was then extracted three times, each time with 1 litre of ethyl acetate, to give M—4. This compound shows an Rf value of 0.45 on thin layer chromatography (TLC plate: Silica Gel Art 5715 manufactured by Merck & Co., Inc.; solvent: a 50:50:3 by volume mixture of benzene, acetone and acetic acid). The extract was washed with a saturated aqueous solution of sodium chloride,			25
	and, after adding an ethereal solution of diazomethane, left standing for 30 minutes. It was then evaporated to dryness under reduced pressure. The residue was dissolved in 10 ml of a 55% v/v aqueous methanol solution, and passed through a column chromatograph (Product of Merck & Co., Inc.; RP—8, Size B). After passing 200 ml of a 55% v/v aqueous methanol solution, it was eluted with a 60% v/v aqueous methanol solution. The first 240 ml of the eluate were discarded, and the next 120 ml were collected. This fraction was evaporated to dryness and the residue was dissolved in 2.5 ml of a 65% v/v aqueous methanol solution and purified by high-performance liquid chromatography (JASCO—Trirotar, column: $\mu$ — Bondapak C <sub>18</sub> ). The portion which showed the fourth peak was separated and the solvent		30 35	
	EXAMPLE 20 Preparation of M—4	bit liver was used in this Example to obtain	M—4 from ML—236B.	
40	were added to one volu	a 1.15% w/v potassium chloride — 10 mN ume of rabbit liver and the mixture was hom or 20 minutes at 9,000 G and the supernata	ogenized. The homogenized mixture	40
45	(b) Cofactor solution		•	45
		$\beta$ -Nicotinamide adenine dinucleotide phosphate (reduced form NADPH)	3 mg	•
		MgCl <sub>2</sub> solution (508 mg/10 ml)	0.1 ml	•
	•	1.15% w/v KCl solution	0.3 ml	
50		0.2 M phosphate buffer solution (pH 7.4)	0.6 mi	50

The above substances wer mixed to a total volume of 1 ml to make the cofactor solution.

40

(c) Reaction solution

80  $\mu$ l of the above enzymatic solution, 20  $\mu$ l f the above cofactor solution and 2  $\mu$ l of a methanol solution of ML—236B w re mixed to make a final concentration of ML—236B of 1 mM. The resulting solution was shaken for 30 minutes at 37 °C. M—4 was formed in the reaction mixture and identified by TLC (the same conditions as in Example 19).

#### **EXAMPLE 21**

Preparation of M-4 sodium salt

2 mg of M—4 methyl ester were dissolved in 1 ml of a 0.1 N aqueous solution of sodium chloride and subjected to hydrolyzation at 30°C for 1 hour. The reaction mixture was washed with 1 ml of chloroform and the resulting aqueous phase was adjusted to pH 8 with 0.1 N hydrochloric acid and passed through a XAD—2 column (about 5 ml). The column was washed with 20 ml of distilled water and the desired product was eluted with 15 ml of 50% v/v aqueous acetone. The acetone was distilled off from the eluate. The residue was confirmed by high-performance liquid chromatography to give a single peak (retention time was 13 minutes, eluted with 40% v/v aqueous methanol at 1 ml/minute). The residue was then 15 lyophilized to give 0.8 mg of M—4 Na salt having the same properties as the product of Example 14.

#### **EXAMPLE 22**

Preparation of M-4 methyl ester

Each of twenty 500 ml. Erlenmeyer flasks containing 100 ml. of a medium having the composition listed below was inoculated with spores of *Mucor hiemalis f. hiemalis IFO*—5834. The inoculum was subjected to shaking culture at 26°C. and 220 rpm. After 4 days, ML—236B was added to a final concentration of 0.05% w/v, and cultivation was conducted at 26°C. and 220 rpm for additional 6 days. The composition of the medium was (percentages are w/v):

	Tap water	balance (pH unadjusted).	
	Corn steep liquor	0.3%	
	Yeast extract	0.1%	
25	Meat extract	0.1%	25
	Peptone	0.2%	
	Glucose	1.0%	

After completion of the cultivation, the filtrate was adjusted to a pH of 3 with trifluoroacetic acid.

The mixture was then extracted three times, each time with 100 ml. of ethyl acetate. There was obtained a fraction containing M—4. M—4 has an Rf value of 0.45 on thin layer chromatography (Plate: Merck Silica gel Art 5715; Solvent: a 50:50:3 by volume mixture of benzene, acetone and acetic acid). The conversion ratio was 90%. This extract was washed with a saturated aqueous solution of sodium chloride, after which there was added an ethereal solution of diazomethane. The resulting mixture was allowed to stand for 30 minutes and then concentrated under reduced pressure to dryness. The residue was placed on a Lobar column (Merck Si 60, size A) and purified with a 1:1 by volume mixture of benzene and ethyl acetate. There were obtained about 600 mg. of M—4 methyl ester, having the same properties as the product of Example 13.

#### 40 EXAMPLE 23

Preparation of M-4 lactone

The procedure described in Example 22 was repeated up to and including washing of the three ethyl acetate extracts with a saturated aqueous solution of sodium chloride. The resulting solution was then evaporated to dryness to give a lactone product. The product was recrystallized from ethyl acetate to give about 560 mg. (56%) of M—4 lactone, having the same properties as the product of Example 1.

#### **EXAMPLE 24**

Preparation of M-4 sodium salt

The procedure described in Example 22 was repeated to give 1.9 litres of the filtrate from the conversion reaction. This was extracted through the etimes, each time with 1 litre of ethyl acetate to giv fractions containing M—4. By immediately transferring these into a 5% w/v aqueous solution of sodium bicarbonate, ther was obtained a fraction containing M—4 sodium salt. Then the M—4 sodium fraction was adjusted with 2N hydrochloric acid to a pH of 7.0 and adsorbed on an HP—20 column (manufactured by Mitsubishi Chemical Industries). Washing with water and elution with 50% v/v aqueous acetone gave a fraction containing M—4 sodium salt, from which ther we re obtained 570

5

mg. (52%) of a fr eze-dried product, having the properties described in Example 14.

### **EXAMPLE 25**

Preparation of M-

The procedure described in Example 22 was repeated, except that the following microorganisms were employed and the conversion to M-4 was as shown by the associated codes:

Conversion to M-4 Microorganism: +4 Mucor hiemalis f. hiemalis IFO---5303 Mucor hiemalis f. hiemalis IFO-8567 Mucor hiemalis f. hiemalis IFO-8449 10 Mucor hiemalis f. hiemalis IFO-8448 10 Mucor hiemalis f. hiemalis IFO-8565 Mucor hiemalis f. hiemalis CBS--117.08 Mucor hiemalis f. hiemalis CBS-109.19 Mucor hiemalis f. hiemalis CBS-200.28 15 +4 Mucor hiemalis f. hiemalis CBS-242.35 15 Mucor hiemalis f. hiemalis CBS-110.19 Mucor hiemalis f. hiemalis CBS-201.65 +4 trace Mucor bacilliformis NRRL-2346 Mucor circinelloides f. circinelloides IFO-4554 +1 20 Mucor circinelloides f. circinelloides IFO-5775 +1 20 Mucor hiemalis f. corticolus NRRL—12473 trace trace Mucor dimorphosporus IFO--4556 +1 Mucor fragilis CBS--236.35 Mucor genevensis IFO-4585 +1 25 trace Mucor globosus NRRL-12474 25 +1 Mucor circinelloides f. griseo-cyanus IFO-4563 trace Mucor heterosporus NRRL-3154 trace Mucor spinescens IAM-6071 trace Mucor chinensis IFO---4772 30 +1 Rhizopus circinans ATCC-1225 30 +1 Rhizopus arrhizus ATCC—11145 +1 Zygorynchus moelleri IFO-4833 Circinella muscae IFO-4457 Circinella rigida NRRL--2341 trace

•	Microor	ganism:		Conversion to M—4	
•	Circinell	a umbellata NRRL—11	713	+1	
	Circinell	a umbellata IFO—445	<b>2</b> ±	+1	
	Circinella	a umbellata IFO—584	2	+1	
5	Actinom	ucor elegans ATCC(	6476	+1	5
	Phycomy	vces blakesleeanus NR	IRL12475	trace	
	Martiere	lla isabellina IFO673	39	trace	
	Gongron	ella butleri IFO8080	)	+1	
	Руспоро	rus coccineus NRRL—	12476	+3	
10	Rhizocto	nia solani NRRL—124	77	+2	10
15	trace = 0.5% or less +1 = 0.5—5% +2 = 5.0—10.0%	·	ons to M—4 have the followi	ng meanings:	15
	described in Example was subjected to sha concentration of 0.00 After completic was adjusted with tri each time with 1 litre	500 ml. Sakaguchi flate 22 was inoculated withing culture at 26°C, a 5% w/v, and cultivation of the cultivation, the fluoroacetic acid to a perfectly acid to	and 220 spm. After 4 days, Non was conducted at 26°C. and econversion reaction mixture. The office was the grant of the office was the grant of the grant office was the grant of the grant of the grant office was the grant of the	ae IFO—4457. The inoculum AL—236B was added to a final and 120 spm for a further 6 days. The was filtered, and the filtrate are extracted three times, —4'. This extract was washed	20 25
•	obtained a lactone pr	oduct. The residue wa acetate system, giving	n chloride and then evaporat is placed on a Lobar column ( g 12 mg. of IsoM—4' lactone	Merck Si 60, size A) and	
30	EXAMPLE 27 Preparation of M—4	lantone			30
35	Each of twenty shown below was inc was subjected to sha concentration of 0.05	500 ml. Erlenmeyer flat oculated with spores of king culture at 26°C. a 5% w/v and cultivation	f Syncephalastrum nigricans	medium having the composition NRRL—12478. The inoculum —236B was added to a final 1220 rpm. 1/v):	35
	Glucose	1%			
	Peptone	0.2%	•		
•	Meat extract	0.1%			
40	Yeast extract	0.1%			40
•	Corn steep liquor	0.3% (pH unadjust d).			
	After completio	n of the cultivation, the	a conversion reaction mixture	was filtered and the filterte	

	benz ne, acetone and acetic acid). This extract was washed with a saturated aqueous solution of	•
	sodium chloride, dried over anhydrous sodium sulphat and subjected to lactonization by adding a	
	catalytic am unt of trifluoroacetic acid. The resulting mixture was then washed with a 5% w/v aqueous	•
5	soluti n of sodium bicarb nate, dehydrated with anhydrous sodium sulphate and vaporated to dryness. The residue was crystallized from ethyl greatet to give about 180 mg, of M—4/ Jactone having the	
٠	The residue was crystallized from ethyl acetate to give about 180 mg. of M—4' lactone having the following physical properties:	5
	1) Nuclear Magnetic Resonance Spectrum:	
	Measured in deuterochloroform at 100 MHz, using tetramethylsilane as the internal standard.	
	δ ppm:	
10	6.01 (1H, doublet);	10
	5.90 (1H, quartet);	
	5.75 (1H, multiplet);	
	5.50 (1H, multiplet); 4.60 (1H, multiplet);	
15	4.25 (1H, multiplet).	15
	2) Ultraviolet Absorption Spectrum (methanol) 1, nm: 230, 237, 245.	
	3) Infrared Absorption Spectrum (KBr) $v \text{ cm}^{-1}$ : 3500, 1720.	
	4) Mass spectrum: M/e: (406(M <sup>+</sup> ), 304, 286.	
20	5) Optical rotation: $[\alpha]_0^{25} = +310.9^{\circ}$ (c = 0.66, methanol).	
	6) Melting point: 141—143°C. 7) Elemental analysis:	20
	77 Elementes attalysis.	
	Calculated : C, 67.95%; H, 8.43%	
	Found : C, 68.05%; H, 8.37%.	
	8) Thin Layer Chromatography:	
25	TLC plate: Merck silica gel Art 5715.	25
	Solvent: Benzene — acetone (1:1 by volume)	
	Rf value 0.64.	
	EXAMPLE 28	
	Preparation of M—4' sodium salt	
30	Following substantially the same cultivation procedures as in Example 27, there was obtained a	30
	conversion reaction mixture.	••
	After completion of the cultivation, the conversion reaction mixture was filtered, and the filtrate was adjusted with trifluoroacetic acid to a pH of 3. It was then extracted three times, each time with 1	
	litre of ethyl acetate to give a fraction containing M—4', which was washed with a saturated aqueous	
35	solution of sodium chloride and immediately thereafter passed into a 5% w/v aqueous solution of	35
	sodium bicarbonate, to give a fraction containing M—4' sodium salt. The agueous layer thus obtained	33
	was adjusted to pH 8.0 with 0.1 N hydrochloric acid and adsorbed on a Diajon HP 20 resin column	
	(manufactured by Mitsubishi Chemical Industries). It was then eluted with 50% v/v aqueous acetone.	
40	The acetone was distilled off, and the residue was freeze-dried to give 1.41 g. of M—4' sodium salt, having the following physical properties:	
	1) Nuclear Magnetic Resonance Spectrum:	40
	Measured in deuterochloroform at 60 MHz, using tetramethylsilane as the internal standard	
	a ppm;	
45	6.00 (1H, doublet); 5.95 (1H, quartet);	
••	5.70 (1H, broad singlet):	45
	5.50 (1H, broad singlet).	
	2) Ultraviolet Absorption Spectrum (methanol) A pm: 230, 238, 246	
	3) Infrared Absorption Spectrum (KBr) v cm <sup>-1</sup> : 3400, 2900, 1680.	
50	EXAMPLE 29	EA
	Preparation of M—4' methyl ester	50
	Following substantially the same cultivation procedures as in Example 27, there was obtained a	
	conversion reaction mixture.	
55	After completion of the cultivation, the conversion mixture was filtered and the filtrate was	
- •	adjusted with trifluoroacetic acid to a pH of 3. It was then extracted three times, each time with 1 litre of ethyl acetate. The combined extracts were washed with a saturated aqueous solution of sodium of sodium.	55
	childred and then an ethereal solution of diazomethane was added thereto. The resulting mixture was	
	anowed to stand for 30 minutes and then concentrated under reduced pressure to dryn, as. The residue	
60	was purified using a Lober column (M. rck RP—8, size A) and a 1.1 by volume mixture of benzano and	
50	acetone as the dev loping s lvent. There were obtained 150 mg. of M—4' methyl st r as a colourless	60

oily substance, having the following properties:

1) Nuclear Magnetic Resonance Spectrum:

Measured in deuterochloroform at 60 MHz, using tetramethylsilan as the internal standard.

6 ppm:

5 6.01 (1H, doublet);

5.90 (1H, quartet);

5.75 (1H, broad singlet);

5.50 (1H, broad singlet);

3.70 (3H, singlet).

2) Ultraviolet Absorption Spectrum (methanol) \(\lambda\_{max} \text{nm}: 230, 238, 246.\)

10

2) Ultraviolet Absorption Spectrum (methanol) λ<sub>max</sub>nm: 230, 238, 246.
 3) Infrared Absorption Spectrum (liquid film) ν cm<sup>-1</sup>: 3400, 1730.
 4) Mass analysis:

Measurement was made after silylation with *N,O*-bis(trimethylsilyl)trifluoroacetamide using a mass spectrometer, type D—300 manufactured by Nippon Electronics.

15 M/e: 654 (M<sup>+</sup>).

CLAIMS

1. Compounds of formula (I):

(in which R represents a group of formula

and ring-closed lactones, salts and esters thereof.

2. Compounds as claimed in Claim 1, having the formula (II):

$$H_3$$
C  $H_3$   $H_0$   $CH_3$ 

25

(in which  $R^1$  represents a hydrogen atom or a  $C_1$ — $C_5$  alkyl gr up) and pharmaceutically acceptable salts of the acid wherein  $R^1$  represents a hydrogen atom.

3. A compound as claimed in Claim 1, having the formula (III):

4. Compounds as claimed in Claim 2, having the configuration shown in formula (IV):

(wherein
 R¹ is as defined in claim 2), and pharmaceutically acceptable salts thereof.
 A compound as claimed in Claim 3, having the configuration shown in formula (V):

6. Compounds as claimed in Claim 2, having the configuration shown in formula (VI):

(wherein

5

15

R<sup>1</sup> is as defined in Claim 2) and pharmaceutically acceptable salts thereof:

7. A compound as claimed in Claim 3, having the configuration shown in formula (VII):

5

10

8. Compounds as claimed in Claim 1, which have the formula (VIII):

(in which  $R^1$  represents a hydrogen atom or a  $C_1$ — $C_5$  alkyl group) and pharmaceutically acceptable salts of the acid wherein  $R^1$  represents a hydrogen atom.

- 9. Compounds as claimed in any one of Claims 2, 4, 6 and 8, wherein R1 represents a hydrogen
- 10. Compounds as claimed in any one of Claims 2, 4, 6 and 8, wherein  $R^1$  represents a  $C_1$ — $C_5$  alkyl gr up.
- 11. Comp unds as claimed in any on of Claims 2, 4, 6 and 8, wherein R<sup>1</sup> repr sents a m thyl group.
  - 12. Comp unds as claimed in any one of Claims 2, 4, 6 and 8, in the form of the alkali metal salts.
  - 13. C mpounds as claimed in Claim 12, in the form of the s dium salt.
  - 14. A compound as claimed in Claim 1, having the formula (IX):

50

15. A process for preparing a compound as claimed in any one of the preceding claims, which comprises enzymatically hydroxylating ML—236B, or ML—236B carboxylic acid or a salt or ester thereof.

16. A process as claimed in Claim 15, wherein the enzymatic hydroxylation is effected by a microorganism of the genus Mucor, Rhizopus, Zygorynchus, Circinella, Actinomucor, Gongronella, Phycomyces, Martierella, Pycnoporus, Rhizoctonia, Absidia, Cunninghamella, Syncephalastrum or Streptomyces, or with a cell-free, enzyme-containing extract from said microorganisms.

17. A process as claimed in Claim 16, wherein said microorganism is:

10 Absidia coerulea Cunninghamella echinulata
Syncephalastrum racemosum
Streptomyces roseochromogenus
Mucor hiemalis f. hiemalis

15 Mucor bacilliformis
Mucoř circinelloides f. circinelloides

Mucor hiemalis f. corticolus

Mucor dimorphosporus

Mucor fragilis

Mucor genevensis

Mucor globosus

Mucor circinelloides f. griseo-cyenus

Mucor heterosporus
Mucor spinescens
25 Rhizopus chinensis
Rhizopus circinens

Rhizopus arrhizus
Zygorynchus moelleri
Circinella muscae
30 Circinella rigida 30

Circinella umbellata Actinomucor elegans Phycomyces blakesleeanus Martierella isabellina

35 Gongronella butleri 35
Pycnoporus coccineus
Rhizoctonie solani

Syncephalastrum nigricans or
Absidia glauca var. paradoxa.

40 18. A process as claimed in Claim 17, wherein said microorganism is:

40 18. A process as claimed in Claim 17, wherein said microorganism is:

Absidia coerulea IFO—4423

Cunninghamella echinulata IFO—4445

Cunninghamella echinulata IFO—4444

Cunninghamella echinulata ATCC—9244
45 Syncephalastrum racemosum IFO—4814 45
Syncephalastrum racemosum IFO—4828

Syncephalastrum recemosum IFO—4828
Streptomyces roseochromogenus IFO—3363
Streptomyces roseochromogenus IFO—3411

50 Mucor hiemalis f. hiemalis IFO-5834

	Mucor hiemalis f. hiemalis IFO5303	-
	Mucor hiemalis f. hiemalis IFO—8567	
	. Mucor hiemalis f. hiemalis IFO—8449	
	Mucor hiemalis f. hiemalis IFO—8448	
	5 Mucor hiemalis f. hiemalis IFO—8565	5
	Mucor hiemalis f. hiemalis CBS—117.08	
	Mucor hiemalis f. hiemalis CBS—109.19	
	Mucor hiemalis f. hiemalis CBS—200.28	
	Mucor hiemalis f. hiemalis CBS-242.35	
10	Mucor hiemalis f. hiemalis CBS—110.19	10
	Mucor hiemalis f, hiemalis CBS—201.65	
	Mucor bacilliformis NRRL—2346	
	Mucor circinelloides f. circinelloides IFO—4554	
	Mucor circinelloides f. circinelloides IFO—5775	
18	Mucor hiemalis f. corticolus NRRL—12473	15
	Mucor dimorphosporus IFO4556	
	Mucor fragilis CBS—236.35	
	Mucor genevensis IFO-4585	
	Mucor globosus NRRL—12474	
20	Mucor circinelloides f. griseo-cyanus IFO4563	20
	Mucor heterosporus NRRL—3154	
	Mucor spinescens IAM6071	
	Rhizopus chinensis IFO—4772	
	Rhizopus circinans ATCC—1225	
25	Rhizopus arrhizus ATCC11145	25
	Zygorynchus moelleri IFO—4833	
	Circinella muscae IFO—4457	
	Circinella rigida NRRL—2341	
	Circinella umbellata NRRL—1713	
30	Circinella umbellata IFO—4452	30
	Circinella umbellata IFO—5842	-
	Phycomyces blakesleeanus NRRL—12475	
	Martierella isabellina IFO—6739	
	Gongronella butleri IFO—8080	
35	Pycnoporus coccineus NRRL—12476	35
	Rhizoctonia solani NRRL—12477	-
	Syncephalastrum nigricans NRRL—12478	
	Syncephalastrum nigricans NRRL—12479	
	Syncephalastrum nigricans NRRL—12480	
40	Absidia glauca var. paradoxa IFO—4431 or	40
	Actinomucor elegans ATCC—6476	
	19. A process as claimed in any one of Claims 15 to 18, wherein there is separated from the	
	reaction mixture one or more of M—4, M—4', IsoM—4, IsoM—4' or a salt, ester or lactone of M—4,	
	M—4' IsoM—4 or IsoM—4'.	
45	20. A process as claimed in Claim 19, wherein said ester is a C <sub>1</sub> —C <sub>5</sub> alkyl ester.	45
	21. A process as claimed in Claim 19, wherein said ester is the methyl ester.	
	22. A process as claimed in Claim 19, wherein said salt is an alkali metal salt.	
	23. A process as claimed in Claim 19, wherein said salt is a sodium salt.	
	24. A process as claimed in Claim 15, wherein said micrographism is:	
50	Absidia coerulea	50
	Cunninghamella echinulata	
	Syncephalastrum racemosum	
	Mucor hiemalis f. hiemalis	
	Mucor bacilliformis	
55.	Mucor circinelloides f. circinelloides	55
	Mucor hiemalis f. corticolus	
	Mucor dimorphosporus	
	Mucor fragilis	
	Mucor genevensis	
60	Mucor globosus	60
	Mucor circinelloides f. griseo-cyanus	
	Mucor heterosporus	
	Mucor spinescens	
	Pycnoporus coccineus	
65	Rhizoctonia solani r	65

	Syncephalastrum nigricans.	
	25. A proc ss as claimed in Claim 15, wherein said microorganism is:	
	Absidia coerulea IFO4423	
	Cunninghamella echinulata IFO—4445	_
5	Cunninghamella echinulata IFO—4444	5
	Cunninghamella echinulata ATCC9244	
	Syncephalastrum racemosum IFO—4814	
	Syncephalastrum racemosum IFO—4828	
	Mucor hiemalis f. hiemalis IFO—5834	
10	Mucor hiemalis f. hiemalis IFO—5303	10
••	Mucor hiemalis f. hiemalis IFO—5304	
	Mucor hiemalis f. hiemalis IFO—8567	
	Mucor hiemalis f. hiemalis IFO—8449	
	Mucor hiemalis f. hiemalis IFO—8448	
15	Mucor hiemalis f. hiemalis 1FO—8565	15
	Mucor hiemalis f. hiemalis CBS—117.08	
	Mucor hiemalis f. hiemalis CBS—109.19	
	Mucor hiemalis f. hiemalis CBS—200.28	
	Mucor hiemalis f. hiemalis CBS—242.35	
20	Mucor hiemalis f. hiemalis CBS—110.19	20
20	Mucor hiemalis f. hiemalis CBS—201.65	
	Mucor bacilliformis NRRL—2346	
	Mucor circinelloides f. circinelloides \text{\$FO}-4554	
	Mucor circinelloides f. circinelloides IFO—5776	
25	Mucor hiemalis f. corticolus NRRL—12473	25
	Mucor dimorphosporus IFO—4556	
•	Mucor fragilis CBS—236.35	
	Mucor genevensis IFO—4585	
	Mucor globosus NRRL—12474	
30	Mucor circinelloides f. griseo-cyanus IFO—4563	30
00	Mucor heterosporus NRRL3154	
	Mucor spinescens IAM—6071	
	Pycnoporus coccineus NRRL—12476	
	Rhizoctonia solani NRRL—12477	
35	Syncephalastrum nigricans NRRL—12478	35
•••	Syncephalastrum nigricans NRRL—12479 or	
	Syncephalastrum nigricans NRRL—12480.	
	26. A process as claimed in Claim 15, wherein said microorganism is:	
	Mucor hiemalis f. hiemalis	•
40	Mucor circinelloides f. circinelloides	40
•••	Mucor fragilis	
	Mucor genevensis	
	Mucor circinelloides f. griseo-cyanus	
	Pycnoporus coccineus or	
45	Rhizoctonia solani.	45
. •	27. A process as claimed in Claim 15, wherein there is prepared a compound of formula (VI):	

(wherein

R¹ represents a hydrogen atom or a C<sub>1</sub>—C<sub>5</sub> alkyl group), a pharmaceutically acceptable salt of the acid wherein R¹ represents a hydrogen atom, or a compound of formula (VII):